## **Environmental Protection Agency**

semichemical chemical recovery combustion unit.

- (d)  $CO_2$ , biogenic  $CO_2$ ,  $CH_4$ , and  $N_2O$  emissions from each kraft or soda pulp mill lime kiln.
- (e)  $CO_2$  emissions from addition of makeup chemicals (CaCO<sub>3</sub>,  $Na_2CO_3$ ) in the chemical recovery areas of chemical pulp mills.
- (f) CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O combustion emissions from each stationary combustion unit. You must calculate and report these emissions under subpart C of this part (General Stationary Fuel Combustion Sources) by following the requirements of subpart C.

## §98.273 Calculating GHG emissions.

(a) For each chemical recovery furnace located at a kraft or soda facility, you must determine CO<sub>2</sub>, biogenic CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O emissions using the procedures in paragraphs (a)(1) through (a)(3) of this section. CH<sub>4</sub> and N<sub>2</sub>O emissions must be calculated as the sum of

emissions from combustion of fossil fuels and combustion of biomass in spent liquor solids.

- (1) Calculate fossil fuel-based  ${\rm CO_2}$  emissions from direct measurement of fossil fuels consumed and default emissions factors according to the Tier 1 methodology for stationary combustion sources in §98.33(a)(1).
- (2) Calculate fossil fuel-based  $CH_4$  and  $N_2O$  emissions from direct measurement of fossil fuels consumed, default HHV, and default emissions factors and convert to metric tons of  $CO_2$  equivalent according to the methodology for stationary combustion sources in  $\S 98.33(c)$ .
- (3) Calculate biogenic  $CO_2$  emissions and emissions of  $CH_4$  and  $N_2O$  from biomass using measured quantities of spent liquor solids fired, site-specific HHV, and default or site-specific emissions factors, according to Equation AA-1 of this section:

$$CO_2$$
,  $CH_4$ , or  $N_2O$  from biomass =  $(0.907.18)*Solids*HHV*EF$  (Eq. AA-1)

Where:

 ${
m CO_2}, {
m CH_4}, {
m or} {
m N_2O}, {
m from \ Biomass} = {
m Biogenic} {
m CO_2}$  emissions or emissions of  ${
m CH_4}$  or  ${
m N_2O}$  from spent liquor solids combustion (metric tons per year).

Solids = Mass of spent liquor solids combusted (short tons per year) determined according to §98.274(b).

HHV = Annual high heat value of the spent liquor solids (mmBtu per kilogram) determined according to §98.274(b).

EF = Default emission factor for CO<sub>2</sub>, CH<sub>4</sub>, or N<sub>2</sub>O, from Table AA-1 of this subpart (kg CO<sub>2</sub>, CH<sub>4</sub>, or N<sub>2</sub>O per mmBtu).

0.90718 = Conversion factor from short tons to metric tons.

(b) For each chemical recovery combustion unit located at a sulfite or stand-alone semichemical facility, you must determine  $CO_2$ ,  $CH_4$ , and  $N_2O$  emissions using the procedures in paragraphs (b)(1) through (b)(4) of this section:

- (1) Calculate fossil CO<sub>2</sub> emissions from fossil fuels from direct measurement of fossil fuels consumed and default emissions factors according to the Tier 1 Calculation Methodology for stationary combustion sources in \$98.33(a)(1).
- (2) Calculate CH<sub>4</sub> and N<sub>2</sub>O emissions from fossil fuels from direct measurement of fossil fuels consumed, default HHV, and default emissions factors and convert to metric tons of CO<sub>2</sub> equivalent according to the methodology for stationary combustion sources in §98.33(c).
- (3) Calculate biogenic CO<sub>2</sub> emissions using measured quantities of spent liquor solids fired and the carbon content of the spent liquor solids, according to Equation AA-2 of this section:

Biogenic 
$$CO_2 = \frac{44}{12} * \text{Solids} * CC * (0.90718)$$
 (Eq. AA-2)

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Where:

Biogenic CO<sub>2</sub> = Annual CO<sub>2</sub> mass emissions for spent liquor solids combustion (metric tons per year).

Solids = Mass of the spent liquor solids combusted (short tons per year) determined according to §98.274(b).

CC = Annual carbon content of the spent liquor solids, determined according to §98.274(b) (percent by weight, expressed as a decimal fraction, e.g., 95% = 0.95).

44/12 = Ratio of molecular weights, CO<sub>2</sub> to carbon.

0.90718 = Conversion from short tons to metric tons.

- (4) Calculate  $CH_4$  and  $N_2O$  emissions from biomass using Equation AA–1 of this section and the default  $CH_4$  and  $N_2O$  emissions factors for kraft facilities in Table AA–1 of this subpart and convert the  $CH_4$  or  $N_2O$  emissions to metric tons of  $CO_2$  equivalent by multiplying each annual  $CH_4$  and  $N_2O$  emissions total by the appropriate global warming potential (GWP) factor from Table A–1 of subpart A of this part.
- (c) For each pulp mill lime kiln located at a kraft or soda facility, you must determine  $\text{CO}_2$ ,  $\text{CH}_4$ , and  $\text{N}_2\text{O}$  emissions using the procedures in paragraphs (c)(1) through (c)(3) of this section:
- (1) Calculate CO<sub>2</sub> emissions from fossil fuel from direct measurement of

fossil fuels consumed and default HHV and default emissions factors, according to the Tier 1 Calculation Methodology for stationary combustion sources in  $\S98.33(a)(1)$ ; use the default HHV listed in Table C–1 of subpart C and the default CO<sub>2</sub> emissions factors listed in Table AA–2 of this subpart.

- (2) Calculate  $CH_4$  and  $N_2O$  emissions from fossil fuel from direct measurement of fossil fuels consumed, default HHV, and default emissions factors and convert to metric tons of  $CO_2$  equivalent according to the methodology for stationary combustion sources in §98.33(c); use the default HHV listed in Table C-1 of subpart C and the default  $CH_4$  and  $N_2O$  emissions factors listed in Table AA-2 of this subpart.
- (3) Biogenic  $CO_2$  emissions from conversion of  $CaCO_3$  to CaO are included in the biogenic  $CO_2$  estimates calculated for the chemical recovery furnace in paragraph (a)(3) of this section.
- (d) For makeup chemical use, you must calculate  $CO_2$  emissions by using direct or indirect measurement of the quantity of chemicals added and ratios of the molecular weights of  $CO_2$  and the makeup chemicals, according to Equation AA-3 of this section:

$$CO_2 = \left[ M_{\left( CaCO_3 \right)} * \frac{44}{100} + M_{\left( Na_2CO_3 \right)} \frac{44}{105.99} \right] * 1000 \ kg/metric \ ton$$
 (Eq. AA-3)

Where:

 $CO_2 = CO_2$  mass emissions from makeup chemicals (kilograms/yr).

M ( $CaCO_3$ ) = Make-up quantity of  $CaCO_3$  used for the reporting year (metric tons per year).

M (NaCO<sub>3</sub>) = Make-up quantity of Na<sub>2</sub>CO<sub>3</sub> used for the reporting year (metric tons per year).

 $44 = Molecular weight of CO_2$ .

 $100 = Molecular weight of CaCO_3$ .

105.99 = Molecular weight of Na<sub>2</sub>CO<sub>3</sub>.

## § 98.274 Monitoring and QA/QC requirements.

(a) Each facility subject to this subpart must quality assure the GHG emissions data according to the applicable requirements in §98.34. All QA/QC

data must be available for inspection upon request.

- (b) Fuel properties needed to perform the calculations in Equations AA-1 and AA-2 of this subpart must be determined according to paragraphs (b)(1) through (b)(3) of this section.
- (1) High heat values of black liquor must be determined no less than annually using T684 om-06 Gross Heating Value of Black Liquor, TAPPI (incorporated by reference, see §98.7). If measurements are performed more frequently than annually, then the high heat value used in Equation AA-1 of this subpart must be based on the average of the representative measurements made during the year.